

Free Energy of Nascent-Chain Folding in the Translocon

James Gumbart,^{†,‡,§} Christophe Chipot,^{‡,||} and Klaus Schulten^{*,†,‡}

[†]Department of Physics, and [‡]Beckman Institute, University of Illinois at Urbana-Champaign, Urbana, Illinois 61801, United States

ABSTRACT: During their synthesis, many water-soluble proteins and nearly all membrane proteins transit through a protein-conducting channel in the membrane, the Sec translocon, from where they are inserted into the lipid bilayer. Increasing evidence indicates that folding of the nascent protein begins already within the ribosomal exit tunnel in a sequence- and environment-dependent fashion. To examine the effects of the translocon on the nascent-chain folding, we have calculated the potential of mean force for α -helix formation of a 10-alanine oligopeptide as a function of its position within the translocon channel. We find that the predominant conformational states, α -helical and extended, reflect those found for the peptide in water. However, the translocon, via its surface properties and its variable diameter, shifts the equilibrium in favor of the α -helical state. Thus, we suggest that the translocon facilitates not only the insertion of membrane proteins into the bilayer but also their folding.



INTRODUCTION

Proteins are synthesized by the ribosome initially as linear chains of amino acids, beginning at the peptidyl transferase center (PTC) deep within the ribosome and extending through its 100-Å-long exit tunnel to the surface.¹ This tunnel is largely cylindrical with a diameter ranging from 10 to 20 Å² and, although originally presumed to be inert, many studies now suggest it plays an active role during protein synthesis.³ For example, peptides that induce translational arrest, such as TnaC, proceed via specific interactions with the walls of the exit tunnel.^{4,5} Furthermore, the activity of macrolides, a broad class of antibiotics that interfere with nascent-chain progression, can be modulated by altering their interaction with the exit tunnel.⁶

The ribosomal exit tunnel also plays a role in nascent protein folding. Numerous studies have demonstrated that the tunnel permits modest folding of the nascent chain into a secondary structure^{7–14} and possibly even some tertiary structure.^{15,16} Recent cryo-electron microscopy maps display α -helical segments within a nascent chain inside the exit tunnel.^{17,18} The exit tunnel can actively modulate folding as well; experiments using 5- and 10-residue poly alanine (polyAla) indicate that specific zones inside the tunnel lower the free energy of the folded state by up to 1.1 kcal/mol.¹⁰

Upon exit of their first few residues, i.e., the signaling element, from the ribosome, nascent proteins destined for secretion or the membrane interact with the signal recognition particle,¹⁹ which recruits a protein-conducting channel, the Sec translocon. The translocon, also known as the SecY and SecE complex in bacteria and eukaryotes, respectively, brings about the insertion of nascent membrane proteins into the lipid bilayer or the translocation of secreted proteins across it.^{20–22} Transfer of the nascent chain to the channel is accomplished by docking of the ribosome to the translocon, which, despite inserting its two cytoplasmic loops into the exit tunnel, does not interfere with the extrusion of the nascent chain.^{23–26} The translocon forms an hourglass-shaped channel with each half, cytoplasmic and periplasmic, tapering to a constrictive hydrophobic pore ring at its

center (see Figure 1).^{27,28} The ribosome-translocon system, thus, forms an almost continuous tunnel from the PTC to the opposite side of a membrane.

For the exit of individual transmembrane (TM) segments into the membrane, the translocon possesses a lateral gate at the interface of two halves of SecY, which, when opened, exposes the nascent chain to the lipid bilayer.^{27,29} It is expected that TM segments will transit through the gate in a preformed helical state, due to the significant free-energy penalty of embedding an exposed peptide backbone into the hydrophobic core.³⁰ Biochemical experiments on the insertion of model peptides as well as Kv1.3, indeed, indicate that TM segments acquire secondary structure already within the ribosome and maintain that structure within the translocon.^{7,9,13} However, not all TM segments form compact states early in their development; for example, both polyVal and glycoporphin A (GpA) remain extended even within the translocon, at least initially.⁷

On the basis of the observed compaction of polyAla and polyLeu inside the ribosomal exit tunnel and translocon and the lack of compaction for polyVal, polyPro, and GpA, Mingarro et al.⁷ suggest that the folding propensity of a nascent chain in those locations generally reflects that for amino acids in an aqueous environment, rather than their membrane-insertion propensity. Yet at least one exception has been observed: the TM segment of VSV-G has a compact structure inside the ribosome, but unfolds upon emergence into the cytoplasm.⁹ In addition, computational studies have demonstrated that confinement, for example, by the ribosome or translocon, can stabilize entropically α -helices.^{31–34} Such confinement may, however, be insufficient to alter significantly the folding pathways of those polypeptides tested in Mingarro et al., hence, leaving open the question of what is the precise effect of the ribosome-translocon system on the folding free-energy landscape of polypeptides.

Received: March 14, 2011

Published: April 27, 2011

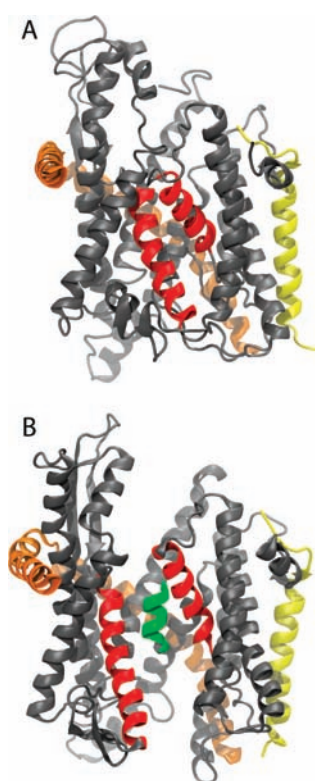


Figure 1. Structure of the SecY complex. The SecY complex is shown from within the membrane plane in a cartoon representation with its three subunits colored gray, orange, and yellow, respectively. The two helices lining the lateral gate are shown in red. (A) SecY complex with a closed lateral gate.²⁷ (B) SecY complex with an opened gate. Ala₁₀ is shown inside the channel as a green helix.

To resolve how confinement in a native environment alters the free energy of folding quantitatively, we determined a two-dimensional potential of mean force (PMF) for helix formation of deca-alanine (Ala₁₀) inside the translocon. The two coordinates used are translocation of the peptide chain along the channel axis z and the average $i, i + 4$ hydrogen-bonding distance ξ along the backbone, that is, the helicity of Ala₁₀. Translocation was restricted over a 30-Å range, from $z = 20$ Å to $z = -10$ Å, where $z = 0$ corresponds roughly to the center of SecY and z is positive in the cytoplasmic half of the channel. Distances between successive hydrogen-bonding pairs were restrained in the second coordinate such that hydrogen bonds form or break concomitantly. The 2D PMF was obtained using adaptive biasing forces (ABF) in NAMD^{35–37} and required about 300 ns of total simulation time (see Methods).

METHODS

Simulations of the translocon began from the crystal structure of the closed state (PDB 1RHZ).²⁷ The channel was embedded in a POPC lipid bilayer, solvated, and ionized with Na⁺ and Cl[−] ions to a strength of 100 mM, using the visualization and analysis program VMD.³⁸ The lateral gate was then opened to a width of 8 Å, in accordance with experimental measurements,³⁹ by applying forces that drove in a near-equilibrium fashion the conformation of SecY toward that of a different structure of SecY with a partially open gate.^{40,41} Finally, Ala₁₀ was inserted in a helical conformation into the center of the channel, which was then equilibrated for 2 ns while the gate and Ala₁₀ were restrained.

MD simulations were carried out using NAMD 2.7³⁶ with the CHARMM force field.^{42,43} The temperature was held at 310 K using Langevin dynamics; the membrane area was held fixed while the pressure normal to the membrane was held at 1 atm using the Langevin piston method.⁴⁴ The equations of motion were integrated employing the RESPA multiple time-step algorithm⁴⁵ with a time step of 2 fs being used for all bonded interactions, 2 fs for short-range nonbonded interactions, and 4 fs for long-range electrostatic interactions. Long-range electrostatic forces were calculated using the particle-mesh Ewald method. Bonds involving hydrogen atoms were constrained to their equilibrium length, employing the Rattle algorithm.⁴⁶

All PMFs were calculated using adaptive biasing forces (ABF), implemented with the *collective variables* module of NAMD 2.7.^{35–37} As noted above, two reaction coordinates were utilized, namely, the position of the center of mass of Ala₁₀ relative to that of the channel and the average hydrogen-bonding distance along the backbone. To improve sampling uniformity, the full reaction-coordinate space was subdivided into multiple regions, which were simulated independently. The resulting data were then combined to produce the PMF.

The *collective variables* module was also employed to enforce a number of restraints to the system. Harmonic potentials were applied between pairs of potential hydrogen-bond partners such that all respective distances were nearly identical; these restraints caused the symmetric formation or deformation of the α -helical state. Additionally, distance restraints were applied to eight pairs of residues across the lateral gate to ensure that its 8-Å opening was maintained. In the periplasmic region of the translocon, that is, for $z < 0$, an additional restraint was required to keep Ala₁₀ from rotating too far off the translocation axis. Forces were applied only when the polar angle of Ala₁₀ was greater than 30°.

Calculation of the channel diameter was performed using HOLE.⁴⁷ The diameter plotted is 2 Å larger than that calculated by HOLE to account for the algorithm's tendency to underestimate the size of irregular volumes.²⁸ The interaction area was calculated over all simulations: for each trajectory frame, the translocation coordinate and the fraction of Ala₁₀ in contact with lipids, or with lipids and hydrophobic residues, were determined. The data were then sorted into 0.2-Å bins in z and averaged over all fractional area values observed for each bin.

RESULTS

As a basis for comparison with the PMF obtained for helix formation in the translocon, we first determined the 1D PMF for folding along the $i, i + 4$ hydrogen-bond length ξ of Ala₁₀ in a purely aqueous environment, shown in Figure 2A. Two folding states are markedly favored, namely, a helical state around $\xi = 3.4$ and a family of extended states with $\xi > 9$ (here and in the following ξ is given in units of Å, though we note that ξ represents the average hydrogen bond length of all $i, i + 4$ pairs of Ala₁₀). A stable intermediate state of slightly higher free energy is also found between the helical and the extended states with two barriers around $\xi = 4.5$ and $\xi = 7.5$ separating the local minima. That both helical and extended states are well represented in the PMF is expected for Ala₁₀. It is known from both theory^{48–50} and experiment^{51,52} that polyAla forms helices in water. Additionally, since Ala₁₀ used here is zwitterionic due to its charged termini, which mimic the properties of an N-terminal signal sequence, the large dipole moment of the extended state contributes to its relatively low free energy.

Many of the features observed in the 1D PMF of folding of Ala₁₀ in water are reproduced in the PMF of folding at different points inside the translocon, shown in Figure 2B. Minima for both α -helical ($\xi = 3.4$) and extended ($\xi > 9$) states are displayed

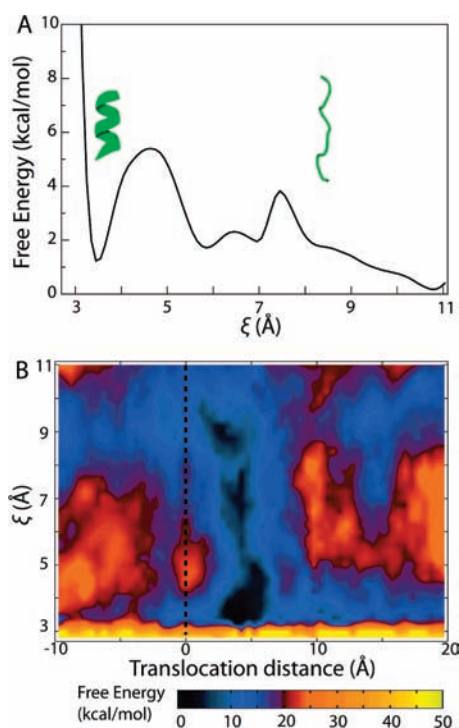


Figure 2. Potential of mean force (PMF) for folding: (A) One-dimensional PMF for Ala₁₀ in water as a function of the concerted-helix-formation coordinate ξ . The two inset proteins (green) represent example configurations for two values of ξ , respectively. (B) two-dimensional PMF for Ala₁₀ in the translocon as a function of the concerted-helix-formation coordinate ξ and translocation distance. The dotted line indicates the position of the translocon's constrictive pore ring.

almost uniformly throughout the channel. In contrast, the barriers separating these states varies with position. Moreover, the free energy of a given Ala₁₀ folding state varies along the channel axis, reaching a global minimum near the pore ring. In particular, the PMF for folding near the center of the translocon, specifically around $z = 5 \text{ \AA}$, appears quite similar to that of Ala₁₀ in a neat aqueous environment. This similarity supports the suggestion of Mingarro et al. that the free-energy landscape of a peptide in the ribosome-translocon system reflects its conformational equilibrium in water.⁷

Projecting the 2D PMF along ξ for different values of z , as shown in Figure 3A, permits a comparison with the 1D PMF of folding of Ala₁₀ in water. At each position within the translocon, clear minima are observed at $\xi = 3.4$, the α -helical state, and at $\xi \geq 9$, the extended states. At nearly all positions, the free-energy difference between the helical state and the minimum of the extended states amounts to between 0 and 3.5 kcal/mol, with the helical state being noticeably more favored on the cytoplasmic side of the channel. Similarly, all PMFs display barriers at intermediate values of ξ ; the heights of these barriers are significantly larger than those observed for folding of Ala₁₀ in water. This increase in the barrier height, ranging from 8 to 15 kcal/mol, suggests that although Ala₁₀ still finds free-energy minima in both helical and extended states, transitions between them are dramatically slowed upon entering the channel, in agreement with coarse-grained simulations of protein folding under confinement.³³ Ala₁₀, thus, most likely retains the folded state it possessed prior to its entrance despite the significantly different environment presented by the translocon.

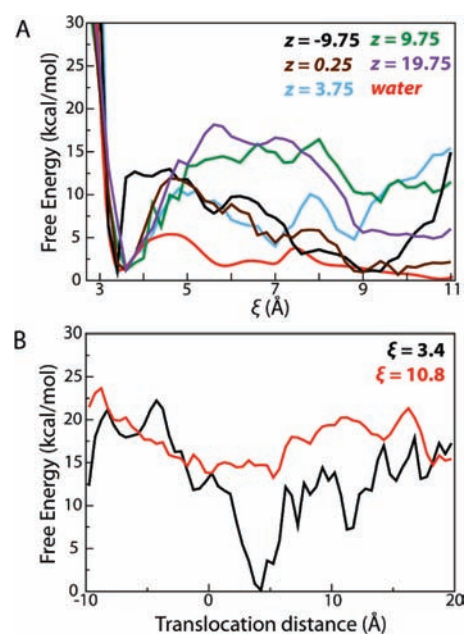


Figure 3. Projections of the 2D PMF in Figure 2B along individual coordinates. (A) Projections of the 2D PMF along different values of z , colored as indicated in the legend, as a function of ξ . The 1D PMF for Ala₁₀ in water is included for comparison. The free energy of the α -helical state was chosen as a reference point for all PMFs. (B) Projections of the 2D PMF for the α -helical ($\xi = 3.4$) and extended ($\xi = 10.8$) states as a function of position in the channel.

The effect of the tapering of the channel to the constrictive pore ring, a hydrophobic gasket-like seal at its center,⁵³ is made most apparent by projecting the 2D PMF along the helical ($\xi = 3.4$) and extended ($\xi = 10.8$) states (see Figure 3B). The free energy of the α -helical state roughly mirrors the diameter of the channel, shown in Figure 4A, reaching a minimum just above the pore ring, whereas the maximally extended state shows little deviation in free energy throughout the channel. Confinement is expected to stabilize the folded state, as observed for Ala₁₀, but also to destabilize the unfolded state, due to a loss of configurational entropy.^{32,33} Yet, on account of the restraints imposed on Ala₁₀ to keep successive hydrogen bonds equidistant, sampling of bent conformations is limited (the propriety of these restraints is discussed below). The entropic effect on folding is, nonetheless, clearly manifested, both in the change in free energy of the helical state across the channel and in the increasing free-energy difference between the extended and the helical states (see Figure 3A).

Beyond purely geometric considerations, the nature of the translocon interior can also affect the free-energy landscape of Ala₁₀. The translocon features a predominantly polar interior on its cytoplasmic side, a hydrophobic constriction in the central pore ring, and a mixed hydrophobic/polar periplasmic side, illustrated in Figure 4A. In addition to interacting with the interior of the translocon, the nascent chain is known to interact with lipids across the lateral gate;^{54–57} in the partially opened translocon model used here (see Methods), the gate opening is $\sim 8 \text{ \AA}$ wide, large enough to allow lipids to interact with Ala₁₀ inside. To characterize the interaction of lipids and Ala₁₀, we measured the fraction of surface area of Ala₁₀ in contact with lipids at different positions within the channel. Shown in Figure 4B, the interaction peaks at 10% near the global free-energy minimum; when the translocon interior is accounted for

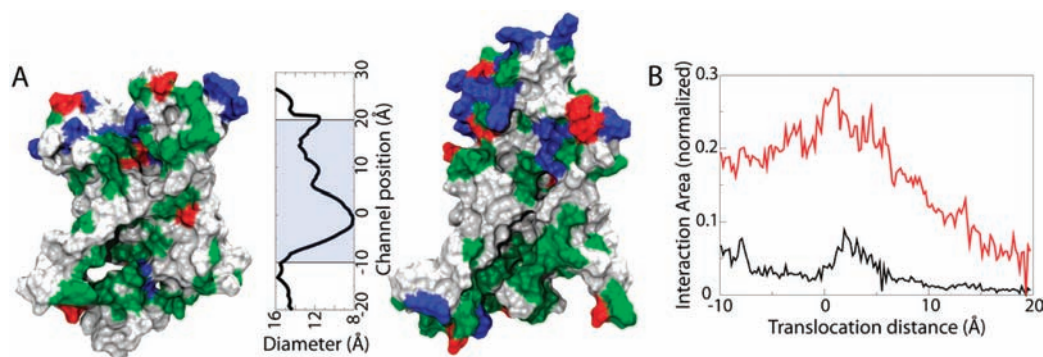


Figure 4. Interaction between Ala₁₀ and the channel. (A) Surface properties of the translocon interior. The N-terminal (left) and C-terminal (right) halves of the translocon interior are shown as molecular surfaces colored according to residue type (nonpolar is white, polar green, basic blue, and acidic red). The channel diameter as a function of channel position is plotted to scale with the channel surfaces. The shaded area indicates the range over which the PMF was calculated. (B) Fraction of the area of Ala₁₀ in contact with lipids (black) and all hydrophobic regions (red) as a function of channel position.

as well, the total fraction of Ala₁₀ interacting with hydrophobic residues or lipids reaches over 25%.

The 1D PMF projections in Figure 3, when correlated with the surface properties of the translocon interior, suggest that predominantly polar surfaces stabilize the helical state of Ala₁₀ relative to the unfolded state, while more hydrophobic surfaces have little effect. The narrow pore ring is an exception, as the helical state is found to have here a significantly lower free energy than the extended states. Given the large contact area of Ala₁₀ with hydrophobic residues and lipids (up to 30% at its peak), direct interactions between the surface and Ala₁₀ play a significant role in determining the free energy of folding. These observed surface-dependent effects on folding of Ala₁₀ are in good agreement with coarse-grained simulations of polyAla inside a nanotube of varying hydrophobicity.⁵⁸ In those simulations, small hydrophobic patches embedded in an otherwise hydrophilic nanotube maximized helical stability, with the effect being most pronounced at large (≥ 20 Å) nanotube diameters.⁵⁸ The confinement of water is also predicted to have an effect on the free-energy landscape of Ala₁₀ folding,^{59,60} for example, by promoting helix formation for polyAla in a predominantly hydrophilic environment, such as the cytoplasmic half of the translocon, as demonstrated in Figure 3B.

CONCLUSIONS

To interpret our simulation results for Ala₁₀ in the translocon, it is necessary to consider assumptions inherent in the setup of the simulations. For example, SecY was modeled in a laterally open state, partially exposing Ala₁₀ to the lipid bilayer. Although direct interactions of the nascent chain and the lipids have been observed experimentally,^{54–57} the degree of lateral gate opening should be considered a flexible property, with an opening of at least 8 Å required to support translocation,³⁹ 11 Å for the intercalation of the signal sequence,⁶¹ and at least 13–14 Å for lateral exit.²⁶ An opening of 8 Å was modeled here, leading to a pore-ring diameter of 8 Å. This diameter hardly accommodates the Ala₁₀ α -helix, which itself is 8-Å wide. The tight conditions faced at the pore ring explains why the free-energy minimum for the α -helix is found almost 4 Å above the ring. Should the gate, and concomitantly the pore ring, be allowed to expand further, the free-energy profile should shift downward along the channel axis, being otherwise unaltered.

While a nascent chain under physiological conditions extends beyond the translocon on both sides, a short 10-residue oligopeptide, Ala₁₀, was used here. This peptide has greater conformational freedom than a typical nascent protein, being able, in principle, to rotate in all directions and to fold back upon itself. To limit such freedom and ensure that Ala₁₀ remains linear, hydrogen bonds along its backbone were enforced to form or break concomitantly, effectively causing α -helix formation to be highly cooperative. Although cooperativity is expected to be only moderate for short oligopeptides, it is proportional to peptide length, making the effect of the imposed restraints akin to embedding Ala₁₀ into a longer nascent chain.^{49,62,63} The exclusion of specific intermediate states, for example, those with partial helicity, from the PMF calculation does not affect the relative free energies of the end states but may serve to raise the barrier between them. Without such exclusions, it may be expected that transitions between the free-energy minima would be enhanced, although the final disposition of Ala₁₀ would remain unaffected.

The PMFs in Figures 2 and 3 demonstrate that the dominant conformational states of Ala₁₀, both helical and extended, are identical to those found in water, in agreement with experimental studies.^{7,14} The translocon does, however, exert, through its surface properties and through its hourglass-like shape, a biasing effect of a few kilocalories per mole (kcal/mol) in favor of the helical state. This helical bias may be necessary for the insertion of TM segments into the membrane, not all of which form stable helices in water alone.⁹ Transition from an extended state to a helical one inside the translocon will, however, be significantly slower than in water as the barrier between the states is raised, by up to 15 kcal/mol in the case of Ala₁₀ (not accounting for the possible effect of excluding partially helical states). A slow, yet inevitable transition to a helical state explains why two populations were observed experimentally for polyVal inside the translocon.⁷ One population was extended, as polyVal would be in water, but the other was more compact, as required for its insertion into the membrane.⁷ Further translocon-based assays focusing on sequences that only weakly favor the extended state over the helical one in water could confirm the modest helical bias predicted here.

Put together, the results presented here indicate that the translocon, specifically the monomeric SecY, is well suited to facilitate the insertion of TM segments. The mostly polar cytoplasmic half-channel functions essentially as an extension

of the ribosomal exit tunnel, allowing whatever secondary structure preformed in the ribosome to be retained in the translocon. Next, for those TM segments that are not strong helix formers outside the membrane, the constrictive and hydrophobic pore ring at the center of the translocon tilts the free-energy landscape to stimulate folding, albeit slowly. The time scale of translation by the ribosome (on the order of 1 s/residue) should, however, provide ample time for TM segments to fold once fully ensconced within the channel. The multifaceted pore ring also helps to define the energetic threshold separating membrane integration versus excretion into the periplasm.^{41,64}

The application of the methods developed here to more realistic polypeptides, including native TM segments that fold in water, as well as those that do not, should further help elucidate the active role played by the translocon in membrane-protein development.

AUTHOR INFORMATION

Corresponding Author

kschulte@ks.uiuc.edu

Present Addresses

⁵Biosciences Division, Argonne National Laboratory, Argonne, Illinois.

¹¹Equipe de Dynamique des Assemblages Membranaires, UMR Centre National de la Recherche Scientifique/UHP 7565, Nancy Université BP 239, Nancy, France.

ACKNOWLEDGMENT

This work was supported by the National Institutes of Health (R01-GM067887 and P41-RR005969) and the National Science Foundation (PHY0822613). The authors acknowledge computer time provided by the Texas Advanced Computing Center, the National Center for Supercomputing Applications through the National Resources Allocation Committee (MCA93S028), the Grand Équipement National de Calcul Intensif and the Centre Informatique National de l'Enseignement Supérieur. J.G. was supported during completion of this report by an Argonne National Laboratory Director's Postdoctoral Fellowship.

REFERENCES

- (1) Nissen, P.; Hansen, J.; Ban, N.; Moore, P. B.; Steitz, T. A. *Science* **2000**, *289*, 920–930.
- (2) Voss, N. R.; Gerstein, M.; Steitz, T. A.; Moore, P. B. *J. Mol. Biol.* **2006**, *360*, 893–906.
- (3) Kramer, G.; Boehringer, D.; Ban, N.; Bukau, B. *Nat. Struct. Mol. Biol.* **2009**, *16*, 589–597.
- (4) Seidelt, B.; Innis, C. A.; Wilson, D. N.; Gartmann, M.; Armache, J.-P.; Villa, E.; Trabuco, L. G.; Becker, T.; Mielke, T.; Schulten, K.; Steitz, T. A.; Beckmann, R. *Science* **2009**, *326*, 1412–1415.
- (5) Trabuco, L. G.; Harrison, C. B.; Schreiner, E.; Schulten, K. *Structure* **2010**, *18*, 627–637.
- (6) Starosta, A. L.; Karpenko, V. V.; Shishkina, A. V.; Mikolajka, A.; Sumbatyan, N. V.; Schluenzen, F.; Korshunova, G. A.; Bogdanov, A. A.; Wilson, D. N. *Chem. Biol.* **2010**, *17*, 504–514.
- (7) Mingarro, I.; Nilsson, I.; Whitley, P.; von Heijne, G. *BMC Cell Biol.* **2000**, *1*, 3.
- (8) Kosolapov, A.; Tu, L.; Wang, J.; Deutsch, C. *Neuron* **2004**, *44*, 295–307.
- (9) Woolhead, C. A.; McCormick, P. J.; Johnson, A. E. *Cell* **2004**, *116*, 725–736.
- (10) Lu, J.; Deutsch, C. *Nat. Struct. Mol. Biol.* **2005**, *12*, 1123–1129.
- (11) Lu, J.; Deutsch, C. *Biochemistry* **2005**, *44*, 8230–8243.
- (12) Lu, J.; Deutsch, C. *J. Mol. Biol.* **2008**, *384*, 73–86.
- (13) Tu, L. W.; Deutsch, C. *J. Mol. Biol.* **2010**, *396*, 1346–1360.
- (14) Peterson, J. H.; Woolhead, C. A.; Bernstein, H. D. *Mol. Microbiol.* **2010**, *78*, 203–217.
- (15) Kosolapov, A.; Deutsch, C. *Nat. Struct. Mol. Biol.* **2009**, *16*, 405–411.
- (16) O'Brien, E. P.; Hsu, S. D.; Christodoulou, J.; Vendruscolo, M.; Dobson, C. M. *J. Am. Chem. Soc.* **2010**, *132*, 16928–16937.
- (17) Bhushan, S.; Gartmann, M.; Halic, M.; Armache, J. P.; Jarasch, A.; Mielke, T.; Berninghausen, O.; Wilson, D. N.; Beckmann, R. *Nat. Struct. Mol. Biol.* **2010**, *17*, 313–317.
- (18) Bhushan, S.; Meyer, H.; Starosta, A. L.; Becker, T.; Mielke, T.; Berninghausen, O.; Sattler, M.; Wilson, D. N.; Beckmann, R. *Mol. Cell* **2010**, *40*, 138–146.
- (19) Halic, M.; Beckmann, R. *Curr. Opin. Struct. Biol.* **2005**, *15*, 116–125.
- (20) Rapoport, T. A. *Nature* **2007**, *450*, 663–669.
- (21) Driessen, A. J. M.; Nouwen, N. *Annu. Rev. Biochem.* **2008**, *77*, 643–667.
- (22) Mandon, E. C.; Trueman, S. F.; Gilmore, R. *Curr. Opin. Cell Biol.* **2009**, *21*, 501–507.
- (23) Ménétret, J. F.; Schaletzky, J.; Clemons, W. M., Jr.; Osborne, A. R.; Skånland, S. S.; Denison, C.; Gygi, S. P.; Kirkpatrick, D. S.; Park, E.; Ludtke, S. J.; Rapoport, T. A.; Akey, C. W. *Mol. Cell* **2007**, *28*, 1083–1092.
- (24) Gumbart, J.; Trabuco, L. G.; Schreiner, E.; Villa, E.; Schulten, K. *Structure* **2009**, *17*, 1453–1464.
- (25) Becker, T.; Bhushan, S.; Jarasch, A.; Armache, J.-P.; Funes, S.; Jossinet, F.; Gumbart, J.; Mielke, T.; Berninghausen, O.; Schulten, K.; Westhof, E.; Gilmore, R.; Mandon, E. C.; Beckmann, R. *Science* **2009**, *326*, 1369–1373.
- (26) Frauenfeld, J.; Gumbart, J.; van der Sluis, E. O.; Funes, S.; Gartmann, M.; Beatrix, B.; Mielke, T.; Berninghausen, O.; Becker, T.; Schulten, K.; Beckmann, R. *Nat. Struct. Mol. Biol.* **2011** in press.
- (27) van den Berg, B.; Clemons, W. M., Jr.; Collinson, I.; Modis, Y.; Hartmann, E.; Harrison, S. C.; Rapoport, T. A. *Nature* **2004**, *427*, 36–44.
- (28) Gumbart, J.; Schulten, K. *J. Gen. Physiol.* **2008**, *132*, 709–719.
- (29) Gumbart, J.; Schulten, K. *Biochemistry* **2007**, *46*, 11147–11157.
- (30) White, S. H.; von Heijne, G. *Annu. Rev. Biophys.* **2008**, *37*, 23–42.
- (31) Zhou, H. X.; Dill, K. A. *Biochemistry* **2001**, *40*, 11289–11293.
- (32) Ziv, G.; Haran, G.; Thirumalai, D. *Proc. Natl. Acad. Sci. U.S.A.* **2005**, *102*, 18956–18961.
- (33) Mittal, J.; Best, R. B. *Proc. Natl. Acad. Sci. U.S.A.* **2008**, *105*, 20233–20238.
- (34) Xu, X.; Cao, D. *Eur. Phys. J. E* **2010**, *32*, 307–318.
- (35) Darve, E.; Pohorille, A. *J. Chem. Phys.* **2001**, *115*, 9169–9183.
- (36) Phillips, J. C.; Braun, R.; Wang, W.; Gumbart, J.; Tajkhorshid, E.; Villa, E.; Chipot, C.; Skeel, R. D.; Kale, L.; Schulten, K. *J. Comput. Chem.* **2005**, *26*, 1781–1802.
- (37) Hénin, J.; Forin, G.; Chipot, C.; Klein, M. L. *J. Chem. Theor. Comp.* **2010**, *6*, 35–47.
- (38) Humphrey, W.; Dalke, A.; Schulten, K. *J. Mol. Graphics* **1996**, *14*, 33–38.
- (39) du Plessis, D. J. F.; Berrelkamp, G.; Nouwen, N.; Driessen, A. J. M. *J. Biol. Chem.* **2009**, *284*, 15805–15814.
- (40) Zimmer, J.; Nam, Y.; Rapoport, T. A. *Nature* **2008**, *455*, 936–943.
- (41) Gumbart, J.; Chipot, C.; Schulten, K. *Proc. Natl. Acad. Sci. U.S.A.* **2011**, *108*, 3596–3601.
- (42) MacKerell, A. D., Jr. *J. Phys. Chem. B* **1998**, *102*, 3586–3616.
- (43) MacKerell, A. D., Jr.; Feig, M.; Brooks, C. L., III. *J. Comput. Chem.* **2004**, *25*, 1400–1415.
- (44) Feller, S. E.; Zhang, Y. H.; Pastor, R. W.; Brooks, B. R. *J. Chem. Phys.* **1995**, *103*, 4613–4621.
- (45) Tuckerman, M.; Berne, B. J.; Martyna, G. J. *J. Chem. Phys.* **1992**, *97*, 1990–2001.

- (46) Andersen, H. C. *J. Chem. Phys.* **1983**, *52*, 24–34.
- (47) Smart, O. S.; Neduvellil, J. G.; Wang, X.; Wallace, B. A.; Sansom, M. S. P. *J. Mol. Graphics* **1996**, *14*, 354–360.
- (48) Tirado-Rives, J.; Maxwell, D. S.; Jorgensen, W. L. *J. Am. Chem. Soc.* **1993**, *115*, 11590–11593.
- (49) Young, W. S.; Brooks, C. L. *J. Mol. Biol.* **1996**, *259*, 560–572.
- (50) Topol, I. A.; Burt, S. K.; Deretey, E.; Tang, T. H.; Perczel, A.; Rashin, A.; Csizmadia, I. G. *J. Am. Chem. Soc.* **2001**, *123*, 6054–6060.
- (51) O’Neil, K. T.; DeGrado, W. F. *Science* **1990**, *250*, 646–651.
- (52) Chakraborty, A.; Kortemme, T.; Baldwin, R. L. *Protein Sci.* **1994**, *3*, 843–852.
- (53) Gumbart, J.; Schulten, K. *Biophys. J.* **2006**, *90*, 2356–2367.
- (54) Martoglio, B.; Hofmann, M. W.; Brunner, J.; Dobberstein, B. *Cell* **1995**, *81*, 207–214.
- (55) Plath, K.; Mothes, W.; Wilkinson, B. M.; Stirling, C. J.; Rapoport, T. A. *Cell* **1998**, *94*, 795–807.
- (56) Urbanus, M. L.; Scotti, P. A.; Fröderberg, L.; Sääf, A.; de Gier, J. W. L.; Brunner, J.; Samuelson, J. C.; Dalbey, R. E.; Oudega, B.; Lührink, J. *EMBO Rep.* **2001**, *2*, 524–529.
- (57) McCormick, P. J.; Miao, Y.; Shao, Y.; Lin, J.; Johnson, A. E. *Mol. Cell* **2003**, *12*, 329–341.
- (58) O’Brien, E. P.; Stan, G.; Thirumalai, D.; Brooks, B. R. *Nano Lett.* **2008**, *8*, 3702–3708.
- (59) Lucent, D.; Vishal, V.; Pande, V. S. *Proc. Natl. Acad. Sci. U.S.A.* **2007**, *104*, 10430–10434.
- (60) Lucent, D.; Snow, C. D.; Aitken, C. E.; Pande, V. S. *PLoS Comput. Biol.* **2010**, *6*, e1000963.
- (61) Egea, P. F.; Stroud, R. M. *Proc. Natl. Acad. Sci. U.S.A.* **2010**, *40*, 17182–17187.
- (62) Williams, S.; Cosgrove, T. P.; Gilmanshin, R.; Fang, K. S.; Callendar, R. H.; Woodruff, W. H.; Dyer, R. B. *Biochemistry* **1996**, *36*, 691–698.
- (63) Klimov, D. K.; Betancourt, M. R.; Thirumalai, D. *Folding Des.* **1998**, *3*, 481–496.
- (64) Junne, T.; Kocik, L.; Spiess, M. *Mol. Biol. Cell* **2010**, *21*, 1662–1670.